

The Effect of Flocculant Type on Settling Properties of Fine Coal Tailings

DM Moyakhe, QP Campbell and E Fosso-Kankeu*

Abstract— Coal - water slurry handling is an integral part of any coal beneficiation process. This is due to the amount of fine coal tailing generated annually as the demand for resources increases. In conjunction with this occurrence and equally important is the water consumption associated with this process. To achieve efficient solid-liquid separation, the correct flocculant choice is crucial in this process. The interaction between the polymer and the slurry is vital in achieving optimum condition. Thus, in this study the influence of flocculant type and dosage on the settling of fine coal tailings was investigated.

Keywords— Coal, Fines, Flocculation, Sedimentation, Tailings.

I. INTRODUCTION

The development of mining has sourced major changes in the past few centuries. The most noticeable change being equipment automation and/or mechanization dominating the underground as well as surface mining industry. The development has brought about increased production volumes, and in contrast, a decrease in selectivity. Thus, an increase in refuse produced [1], with possible impact on the environment such as acid mine drainage formation [2 – 6].

The coal mining industry is similar; with increased volumes of coal produced and mineral matter refuse to report to the coal preparation plant. This resulted in the generation of a significant amount of wastewater with fine and colloidal particle tailings, composed of organic and inorganic matter which may often be reported in surface waters [7 – 16]. As a result, solid-liquid processes are thus necessary to manage wastewater slurry, and tailings disposed to recover the plant water [17].

Fine coal tailings handling is considered as the most challenging and exorbitant operation [18]. Thus, an appropriate dewatering method is essential. In practice, the prevalent method used is sedimentation with aid of flocculation in a thickener, using polymeric flocculants. The method has two significant parameters that evaluate flocculation performance [19]. The first parameter is the low water turbidity, since water

recovered at the thickener overflow is recycled to the plant. The suspended colloidal particles in the recycled water should be at the lowest level, eliminating any negative implications to the process. The second parameter, settling rate of the flocculated particles, has a direct effect on thickener capacity as well as flocculant performance. Thus, the challenge is achieving the lowest water turbidity level while maintaining high settling rate. Furthermore, optimizing polymeric flocculant properties and wastewater sludge pH is crucial in realizing preferred water clarity and settling rate.

The complex nature of flocculation process requires a precise determination of various properties of the wastewater sludge, this includes physical, chemical and electro-kinetic properties of the solid matter in the sludge. Moreover, the efficiency of the process is dependent on the use of accurate chemical reagents – coagulants, flocculants, etc. – as well as the addition thereof [1].

The main objective of this study is to examine the physical, and mineralogical characteristics of fine coal tailings, investigate the flocculation performance of various types of flocculants and select the most appropriate. The type of flocculants based on their physical characteristics such as the molecular weight and the charge density was varied to study the flocculation mechanism of multi-component fine coal tailings slurry. A secondary objective is to determine the type of polymer current being used at the processing plant.

II. EXPERIMENTAL

A. Material

The coal slurry samples utilized in the experiments, were collected from a thickener feeding stream in a coal processing plant in Mpumalanga, South Africa. The representative samples were then transported to the laboratory in sample containers – 200 liter drums – that operated as a stirring tanks as well. To obtain homogenous slurry samples, the samples underwent sufficient stirring.

The solid concentration, pH, and density of the slurry were determined. Using the volumetric method, water hardness as well as the Ca²⁺ and Mg²⁺ concentrations were determined. By means of X-ray fluorescence (XRF) and X-ray diffraction (XRD), the chemical composition and mineral composition of the solid tailings were determined respectively.

High molecular weight polymer flocculants – provided by SENMIN, a division of AECI Mining Solutions Limited – were used in the settling tests. The generic – SENMIN approved – properties of the flocculants are given in TABLE I. A

DM Moyakhe is with the School of Chemical and Minerals Engineering, North West University, Potchefstroom Campus, Private Bag X6001 2520, South Africa.

QP Campbell was with School of Chemical and Minerals Engineering, North West University, Potchefstroom Campus, Private Bag X6001 2520, South Africa.

E Fosso-Kankeu*, School of Chemical and Minerals Engineering, North West University, Potchefstroom Campus, Private Bag X6001 2520, South Africa.

homogenous solution of the flocculant – concentration 500mg/l – was prepared with process water collected from the plant prior to each testing. The pH of the slurry was adjusted using hydrochloric acid to decrease the pH, or lime solution to increase the pH.

TABLE I: CHARACTERISTICS OF FLOCCULANTS

Commercial name	Name	Type	Charge density	Molecular weight	Supplier
-	Floc A	-	-	-	Plant
SENFLOC 5310	Floc B	Non-ionic	-	High	SENMIN
SENFLOC 5330	Floc C	Non-ionic	-	Medium	SENMIN
SENFLOC 5150	Floc D	Anionic	Low	High	SENMIN

B. Method

The flocculation experiments were carried out using batch settling jar tests. For each test, 1 liter of the original slurry – with a solid concentration of 4.71% – was transferred in a glass jar. The desired flocculant solution added to the testing jar. A lid was placed on the jar and mixing

The interface between the water and height of slurry the slurry as a function of time was recorded. From the recorded height versus time graph, the settling rate of the flocculated suspension is calculated. At time 15 minutes, the supernatant water was sampled to measure the turbidity of the clarified water.

III. RESULTS AND DISCUSSION

A. Characterization of Fine Coal Tailings

The XRD results, represented in TABLE II, indicate that the main minerals present in the coal tailings are Quartz, kaolinite, anatase, troilite, and magnetite in the order of decreasing quantity. The chemical composition depicted in Fig 1, as determined by XRF, shows the chemical composition. From the XRF results, the presence of Al₂O₃ and Fe₂O₃ validates the presence of clay minerals in the tailings [20]. The remaining fraction of the clay mineral – 56.58% SiO₂ – is attributed to quartz. The presence of Sulphur is validated by the Sulphur containing mineral troilite, with few carbonate minerals present in tailings, as confirmed by low MgO and CaO percentages. TABLE II, further shows a low slurry pH.

TABLE II: CHARACTERIZATION RESULTS OF THE COAL TAILINGS

Slurry	
pH	3.80
Solid content (%)	4.71
SG	1.03
Conductivity (mS)	3.10
Mg ²⁺ concentration (mg/L)	200.43
Ca ²⁺ concentration (mg/L)	561.01
Solids	
Main components	Quartz, kaolinite, anatase troilite and magnetite
Process water	
pH	4.20
Conductivity (mS)	2.01
Mg ²⁺ concentration (mg/L)	170.05
Ca ²⁺ concentration (mg/L)	298.02

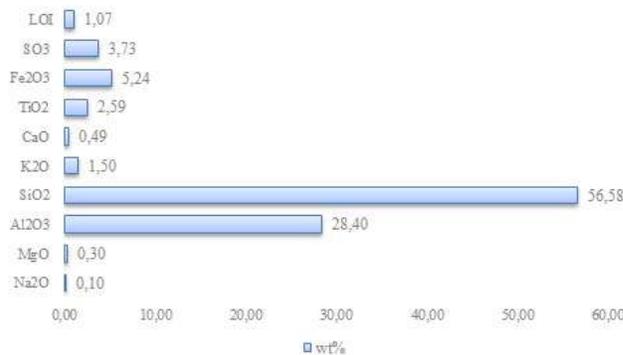


Fig 1: X-ray fluorescence results for the coal tailings

The characteristic of water in the slurry has a significant effect on the surface charge of the quartz mineral and coal that is present in the tailings. At low pH, the surface charge on the clay and coal particles is said to be negative. Thus, low anionic to non-ionic flocculants are the most effective for low pH slurry[21, 22]

B. Flocculation Experiments

The settling tests were performed with the aid of different types of flocculants varying the desired dosages as presented in TABLE I. Fig 2, reports the effect of polymer type and dosage on the settling rate of the coal tailings. Floc A, presented in TABLE II, is the polymer currently employed at the coal preparation plant, and no information was given for this polymer. Also in Fig 2, Floc D - low anionic polymer, the results show an increase in settling rate with an increase in dosage until a maximum was reached. In this case a maximum of 305 mm/min at 97 g/ton. Floc B – C, the non-ionic polymers, however reached higher settling rates at higher polymer dosages compared to the low charged polymer. It is assumed that conformation of the polymer chain is the main contributor to the difference observed between the low anionic and non-ionic polymers [23]. Due to the low pH of the slurry, the low anionic flocculant is unable to extend the chains effectively due to the charge in the functional groups of the polymer. Thus, the bridging mechanism used by this polymer to form bigger flocs is retarded. Non-ionic flocculants relies on the hydrogen bonding for effective floc formation, since the surface charge on the clay minerals, kaolinite and quartz, is negative[1]. The hydrogen bonding aids the formation for floc and hence achieves high settling rates. Furthermore, a difference is detected between Floc B and Floc C in Fig 2. The settling rate of Floc B and C, at dosage 193 g/ton, is 1110 mm/min and 745 mm/min respectively. This difference is due to the difference of molecular mass, the settling rate increases with an increase in mass for the same dosage.

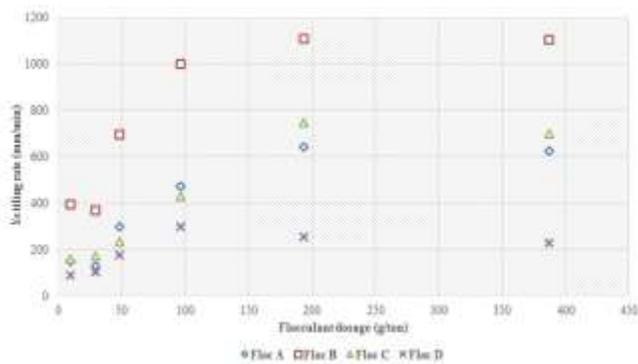


Fig 2: Settling rate versus flocculant dosage

The performance of Floc A, is similar to that of Floc C. This means that Floc A behaves as a non-ionic polymer. Fig 3, shows how the turbidity of the suspension varied with the change of polymer dosage. At low polymer dosage, colloidal particles destabilization is not sufficient [24]. Thus not many colloidal particles forms floc sizes required for settling. As a result, high turbidity values were recorded at low flocculant dosages. As the dosage increase, the lowest turbidity value is reached for all four polymers. Flocs A, C, and D reach lowest turbidity values at dosage 97 g/ton, while for Floc B, the lowest turbidity value is achieved at dosage 48 g/ton. Furthermore, at high dosages the high settling rates results in the dispersion of smaller floc. Thus, surface charges of the colloidal particles particularly the kaolinite in the system, experiences repulsion forces stabilizing the colloidal suspension [22].

IV. CONCLUSION

The characterization of tailings from the coal processing plant showed that the clays kaolinite and quartz to the main constituents of the tailings. In conjunction with low slurry pH, it was showed that choosing the correct polymer for solid-liquid separation is a complex procedure.

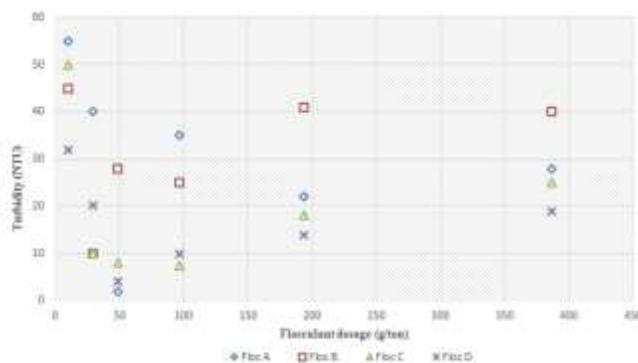


Fig 3: Turbidity versus flocculant dosage

Various factors need to be considered to in order to make informed choice. Different properties of polymers yielded different results. Although, it was shown that for the slurry with a low pH, a non-ionic flocculant is most effective due to the electrostatic forces that results from the negative charge on the surface of the particles.

It was also clear that the molecular mass of the flocculants plays an important role. Despite the lack of information

regarding the properties of Floc A, which is the polymer from the processing plant, performance similar to a non-ionic polymer was observed. The optimum dosage for Floc A, B, and C was shown to be 193 g/ton and 97 g/ton for Floc D. For the system, if settling is the most important factor, then dosage of Floc B at 193 g/ton will be recommended. However, if water turbidity is the most important parameter, it will be advise to dose with Floc A at 48 g/ton.

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The corresponding author is currently an Associate Professor in the School of Chemical and Minerals Engineering at the North-West University (Potchefstroom). He is an NRF rated researcher who has published journal articles, book chapters and book.

Prof Elvis Fosso-Kankeu has been the recipient of several merit awards.